



# **30 Years of Experience Applied to Parking Lots**

**by**

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# WHY AM I PRESENTING THIS TOPIC?

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I have been studying the use of Liquid De-icers for Snow and Ice Control for 27 years ! My County has been **using liquids for 35 years!**

*And*

✗ We have common enemies !!

*What ???*

✗ Snow

✗ Ice

✗ Traffic (In your case pedestrians, as well as vehicles.)



# A LITTLE INFO ABOUT THE COUNTY

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- × 1400 lane km of rural highways
- × 4 maintenance -2 works garages each
- × Winter Budget of \$4.5 Million
- × Equip. replacement budget of \$1.5 Million
- × use 5,000 t of white salt
- × use 5,000 t of treated orange salt (Thawrox)
- × use 30,000 t of sand
- × using liquid de-icers since mid 70's
- × used over 7,000,000 ltrs of liquid in the last 6 years

# A LITTLE ABOUT OUR EQUIPMENT

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- ✖ computerized pre-wet control since 1992
- ✖ 26 combination sander/plows all with computerized spreader controls + pre-wet
- ✖ 24 of those are 3 in 1 units
- ✖ direct liquid units
  - + 2 – 8,000 litre on tandem truck c/w plow and wing (3 lanes)
  - + 1 – 2,000 litre on 3 tonne pick-up (1 lane)

# WHAT AM I GOING TO TALK ABOUT?

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**LOTS !!**

- × EC Code of Practice**
- × Liquids**
- × Anti-icing**
- × Training**

# CURRENTLY

- ✘ Environment Canada formed a committee called the Multi-stakeholders Working Group on Salt Management who produced
    - + A Code of Practice for the Environmental Management of Road Salts
  - ✘ Who are affected?
    - + Any municipality which uses 500 tonnes of salt and /or
    - + Has vulnerable areas (area adversely impacted by the use of road salts)
- and in the future it may include those that
- + Maintain Parking Lots



# 2001 Environment Canada Report

**“Road salts are entering the environment in large amounts and are posing a risk to plants, animals, birds, fish, lake and stream ecosystems and groundwater.”**



# WHY IS THAT?

- ✗ Each year Canada releases 5.2 MILLION TONNES of chloride salts into the environment from road salt
- ✗ Of that total, Ontario is responsible for 1.9 MILLION TONNES
- ✗ Parking lots account for ??% of that total It was 1<sup>st</sup> believed that this was 5-10% now 25-35%





# **SO WHERE DO WE START?**

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**We'll start with liquids.**

**But First:**

**lets look at some Chemistry Terms that  
relate to De-icing products.**

# GENERAL CHEMISTRY TERMS

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- × **Effective Temperature** - Is the lowest temperature in which the cost of the application is justified by the results obtained.
- × **Eutectic Temperature** - The temperature at which a liquid freezes in a concentrated solution in laboratory conditions.
- × **Viscosity** - The quality or property of a fluid that causes it to resist flow.

# GENERAL CHEMISTRY TERMS

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- × **Exothermic** - A chemical reaction that gives off heat. Magnesium Chloride ( $\text{Mg Cl}_2$ ) gives off heat which makes it less reliant on air temperature and thus can work at lower range.
- × **Endothermic** - A chemical reaction that absorbs heat. Salt absorbs heat but in doing so reduces it's effective operating temperature.



# LET'S GET SOMETHING STRAIGHT!!

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**LIQUIDS are :**

- ✗ **the most misunderstood**

**and,**

- ✗ **the most underutilized**

**and,**

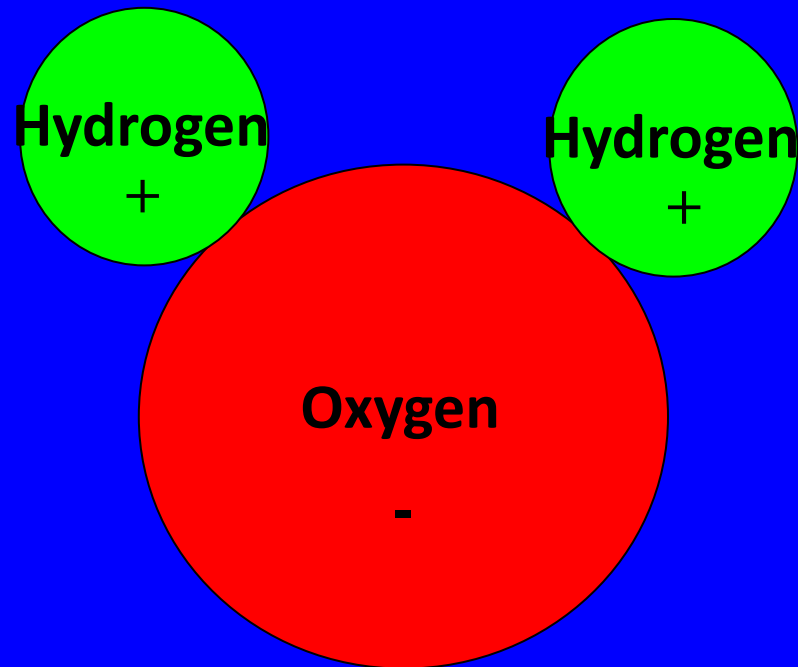
- ✗ **the most misused tool available to you to fight winter !**

# SO: How Does Water Freeze?

**There are three basic states of water**

- **Vapour (gaseous)**
- **Liquid**
- **Solid (ice crystals)**

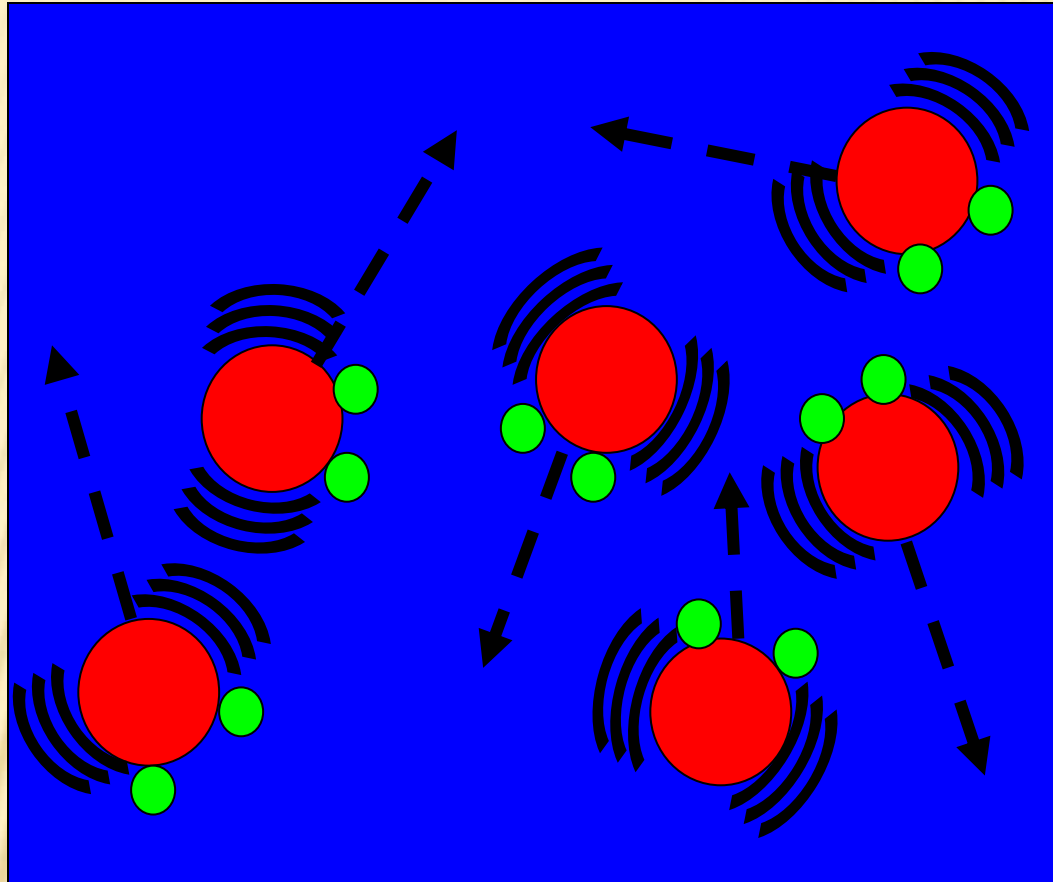
# The Water Molecule



Water (H<sup>2</sup>O) consists of 1 oxygen atom and 2 hydrogen atoms. The O is negatively charged and H are positive

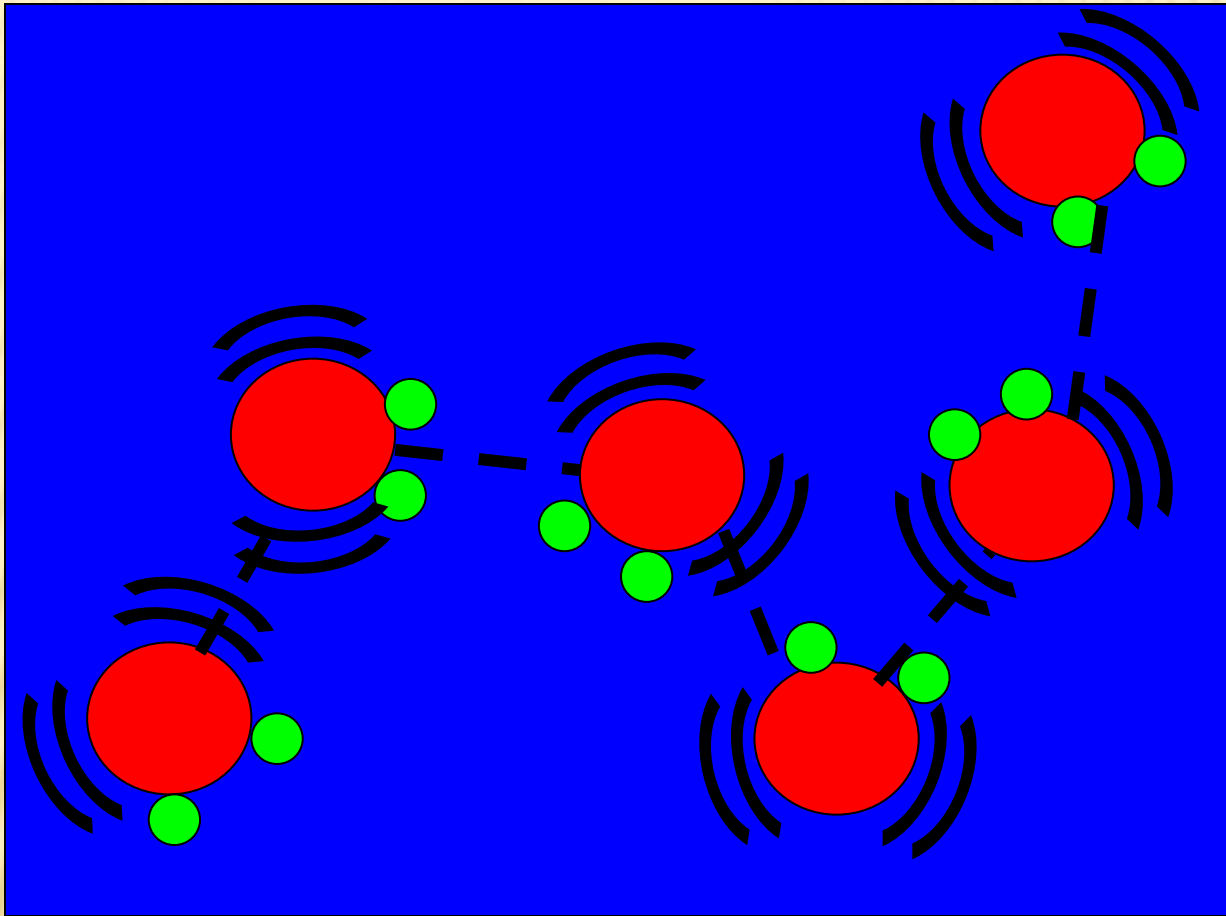


# Vapour



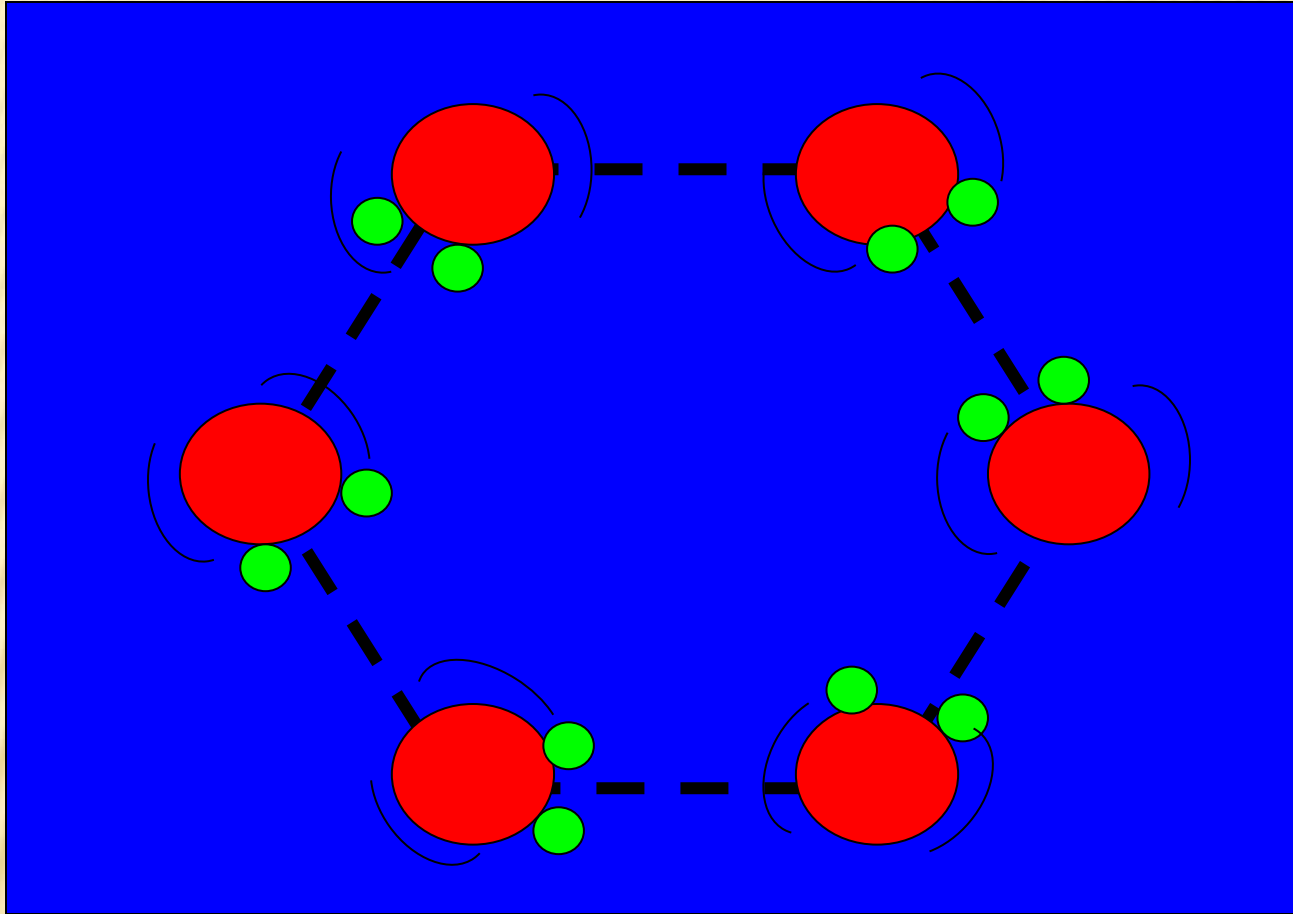
**Water vapour molecules form poor connections as they move about rapidly. Heat causes water molecules to break down.**

# Liquid (Water)



**Water molecules form loose attractions to each other until heated or cooled, which changes their attraction accordingly.**

# Ice Crystals

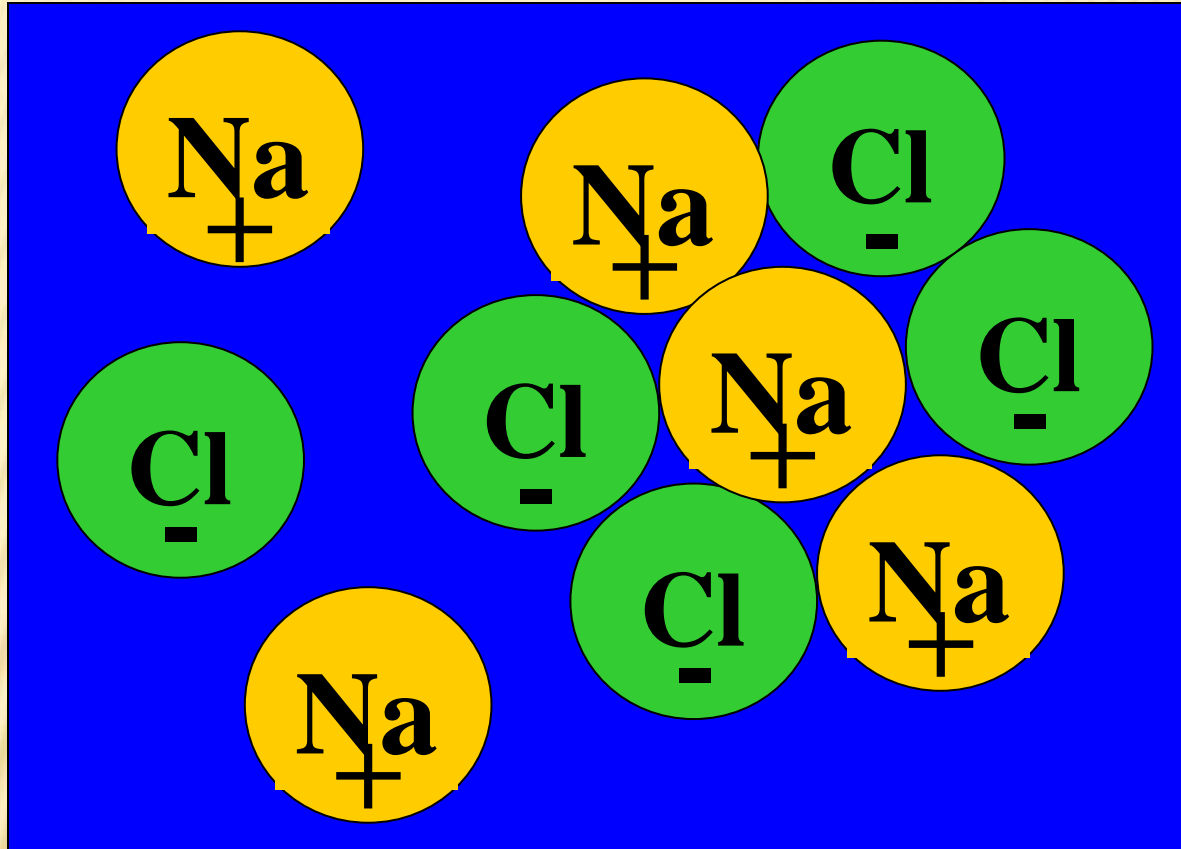


**When water molecules are cooled they move slowly and attach to each other forming these 6 sided (octagonal) ice crystals.**



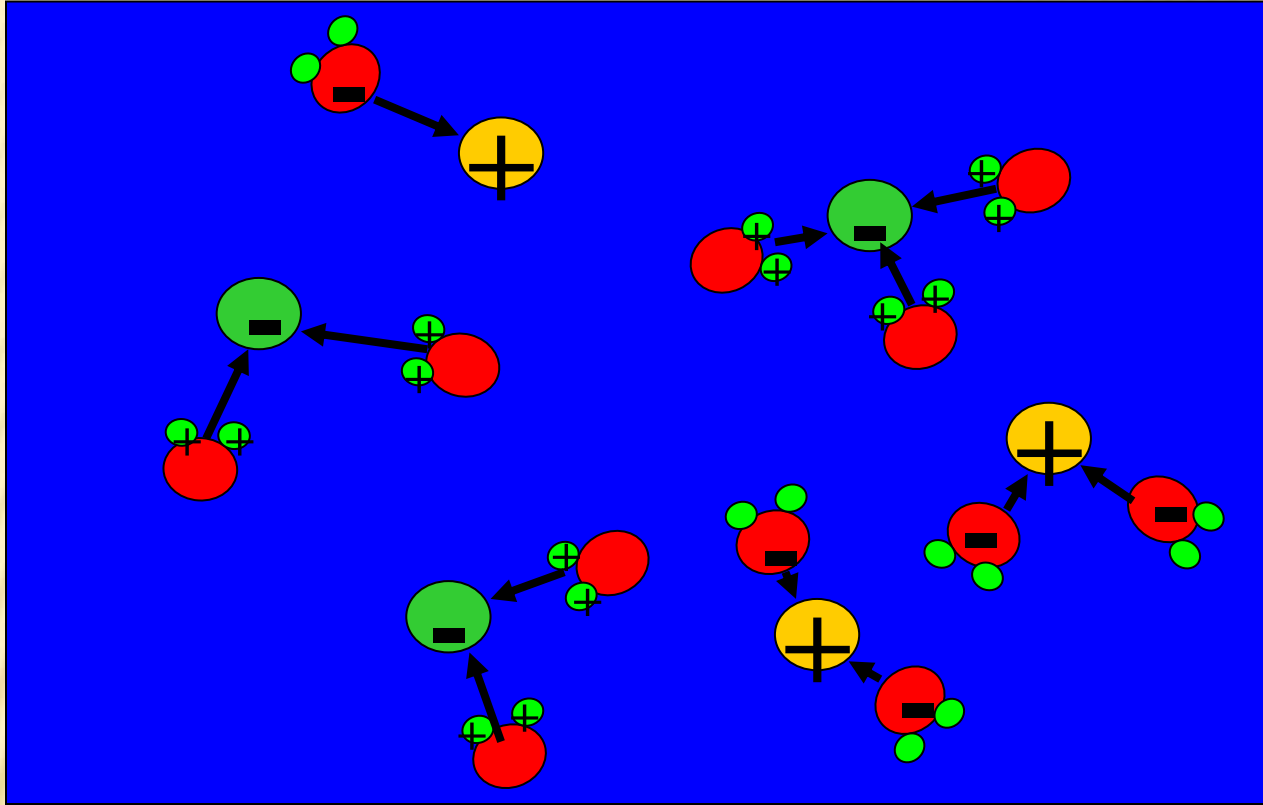
**Freeze-Point-Depressant** – Is a material that has been used to prevent precipitation and condensation from freezing as quickly as it otherwise would have. (It is the ability to lower the freezing point of water.)

# Salt Solution



Salt (sodium chloride) NaCl is made up of 1 positively charged sodium (Na) atom and 1 negatively charged chloride (Cl) atom. Salt is a freeze-point-depressant.

# Melting



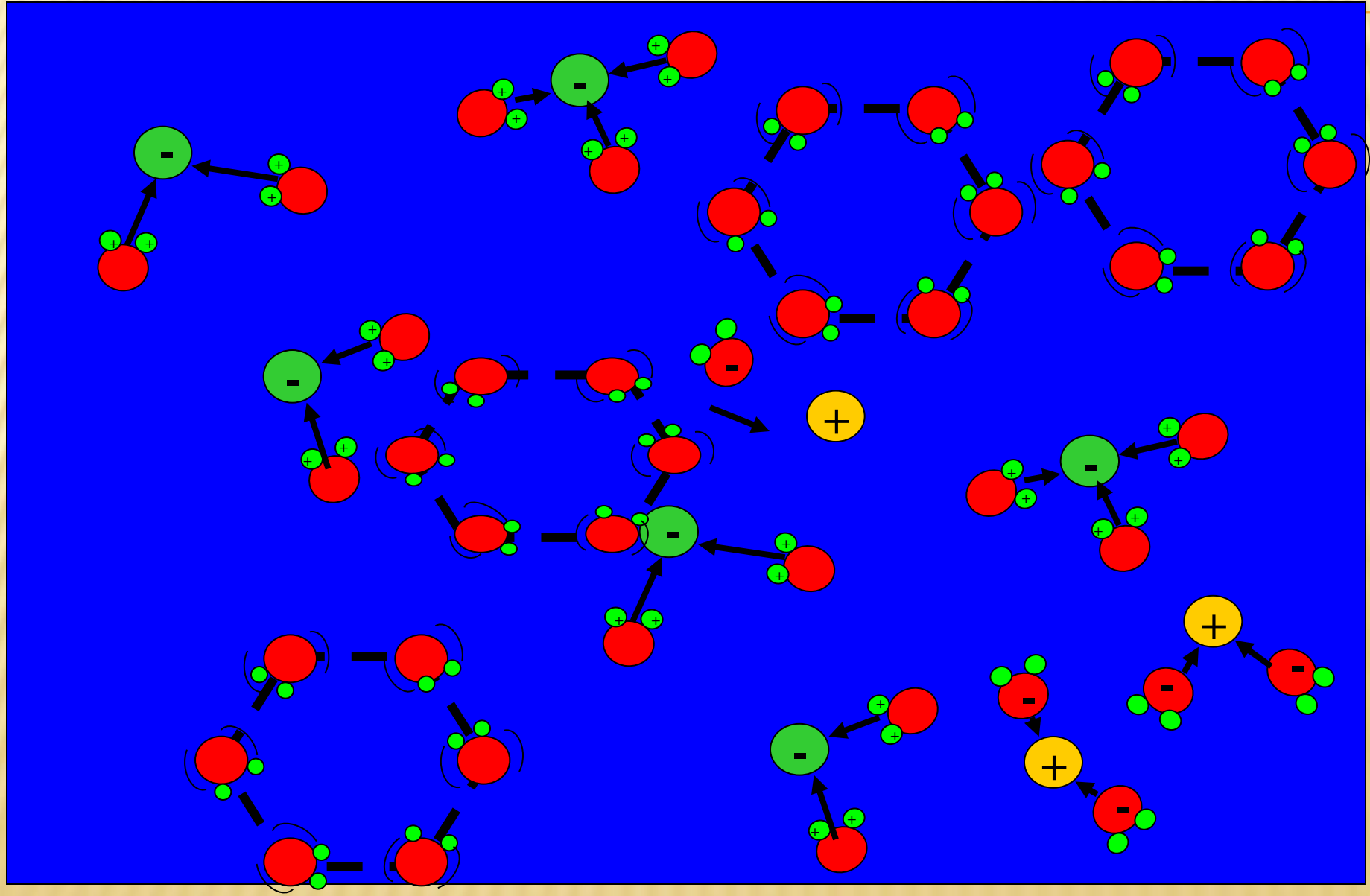
**When salt is applied to snow/ice the positively charged sodium atom attracts the negatively charged oxygen atom and likewise the negatively charged chloride atom attracts the positively charged hydrogen atoms, causing the 6 sided ice crystal to break apart.**



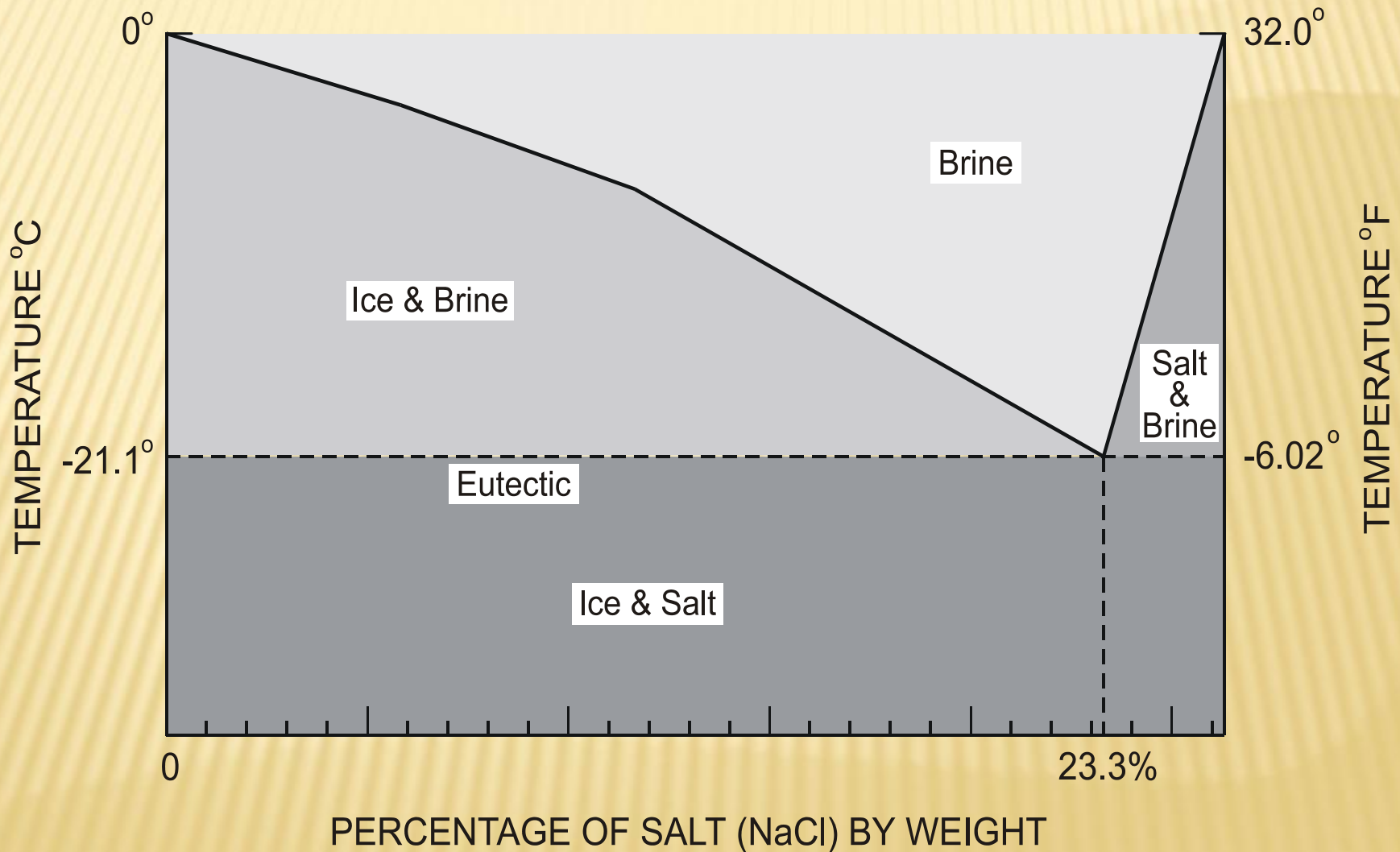
**As the salt solution dilutes it loses the magnetic attraction between the different chemicals and ice crystals re-form (refreeze).**

**The usable working temp. of salt is  $-8^{\circ}\text{C}$  that is why we use a liquid deicer with a usable working temp. of  $-35^{\circ}\text{C}$ .**

# Dilution and Refreeze



# THE EUTECTIC DIAGRAM FOR NaCl





# ***WHY USE LIQUIDS ?***

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- ✖ Once wetted with **Liquid *FPD*:**
  - + **Salt works faster**
  - + **Increases penetration**
  - + **Less blow-off**
  - + **Less bounce and scatter**
  - + **Increased level of service**
  - + **Environmental Savings**

# HOW CAN I USE LIQUIDS ?

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**There are three basic ways to use liquids:**

- 1. Pre-mix**
- 2. Pre-wet**
- 3. Direct application to the road (anti icing and de-icing)**

# WHAT IS PRE-MIXING ?

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*Adding a liquid deicer to your salt before you stockpile it.*

## **Advantages**

- ✖ Better mixing then pre-wet
- ✖ No equipment changes
- ✖ Up to 25% less salt



# MORE ON PRE-MIXING

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## Disadvantages

- ✗ Hard to co-ordinate material, stacker, pug mill, loader, and liquid without rain.
- ✗ Usually one fill, part fills not economical
- ✗ Typically salt is a “just in time” product, not stockpiled **New “Thawrox”**
- ✗ Leaching can occur

# PRE-MIXING SALT

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# WHAT IS PRE-WETTING?

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*The application of a liquid deicer just before the salt hits the road.*

## Advantages

- ✗ More liquid can be added than in stockpile
- ✗ Increase or decrease liquid application as temp. change
- ✗ 25 % savings in salt used



# MORE ON PRE-WETTING?

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## Disadvantages

- ✗ Computerized/liquid controls costly
- ✗ Onboard liquid tanks and pumps
- ✗ Site storage of liquids
- ✗ More costly in short term but will pay for themselves during the life of the truck
- ✗ Additional training required for:
  - + Liquids
  - + Controllers

# PREWETTING EQUIPMENT



# WHAT IS ANTI-ICING?

*a concept in which a liquid, or a pre-wetted salt, is spread directly onto the pavement before the storm begins.*

## Advantages

- × Prevents bonding of snow and /or ice to road  
(Studies have shown it costs 10 times more to break the bond than to prevent the bond.)
- × Faster cleanup
- × Applied in better weather
- × Doesn't get plowed off
- × Equipment is fairly easy to make



# MORE ON ANTI-ICING

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## Disadvantages

- ✗ Need better weather forecasting
- ✗ Larger liquid storage tanks



# ANTI-ICING



# ANTI-ICING UNITS





# ANTI-ICING



This side sprayed

This side unsprayed

# **SAME ROAD DIFFERENT TREATMENTS**



**Traditional Treatment**



**Liquid Treatment**



# DE-ICING

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*an operation where a deicer is applied to an accumulation of snow, ice, or frost that is bonded to the pavement surface.*

## Advantages of liquids verses solids

- ✗ Works much faster than straight salt
- ✗ Residual liquid becomes an anti-icing operation
- ✗ Improved level of services

## Disadvantages

- ✗ Re-active rather than pro-active

# DE-ICING



# **SO FACTS ABOUT LIQUIDS ?**

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- **Liquids are already in solution**
  - **They act more quickly – a strength**
  - **They dilute more quickly – a weakness**
- **Never use with rain (except for pre-wetting)**
- **Never put on top of compact (unless you are using a very hot product)**
- **More is not better, especially early in the season**



# **CORROSION CONCERNS**

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**Because liquids are already in solution:**

- ✗ corrosion quicker than solid chemicals. (if using a liquid that is corrosive)**
- ✗ Major and valid concerns about equipment and infrastructure damage (depends on the liquid)**
- ✗ Some concerns seem to be more perception than reality (misunderstanding)**



# CHEMICAL SLICKNESS

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- ✗ Know what you are applying
- ✗ Equipment is calibrated
- ✗ Let the temperature drop before you make that first application
- ✗ Be especially careful after long dry spells
- ✗ Reduce application rates early in the winter
  - Half rate for first application

# CONCLUSIONS ON LIQUIDS

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- ✗ Liquids are not for melting snow and ice
  - + use them to prevent and/or break the bond instead
- ✗ Road surface temperature not air temperature
- ✗ We need liquids under the snow – anti-icing gets them there faster
- ✗ If we are going to anti-ice, know the limitations as well as the benefits

# MORE CONCLUSIONS

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- ✗ Use liquids that mix well and stay mixed
- ✗ Effective temperature more important than Eutectic.
- ✗ The more Viscous the better
- ✗ Exothermic liquids better than Endothermic
- ✗ Have liquid specific training

# CALIBRATING SPREADERS

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- ✖ Ensure accurate discharge of material
- ✖ Recalibrate at least annually and periodically as required
- ✖ Gate setting is tied to calibration
- ✖ Ensure gate setting is matched to material being applied

<http://www.saltinstitute.org/snowfighting/6-calib.html>



# AND FINALLY TRAINING

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- ✖ Lots out there !
- ✖ OGRA has the WMOT Winter Maintenance Operations Training for Operators (2 days)
- ✖ Winter Maintenance Operations Training for Supervisors and Patrollers (2 days)
- ✖ ½ day Refresher of each of those
- ✖ Snow School
- ✖ TAC , Salt Institute

# ANY QUESTIONS?

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